

Assessing long-term change in forest soils fertility by mean of stable Sr isotope dendrochemistry

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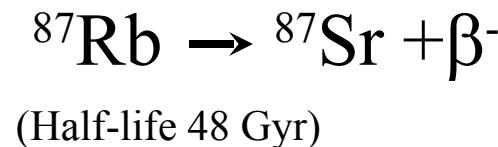
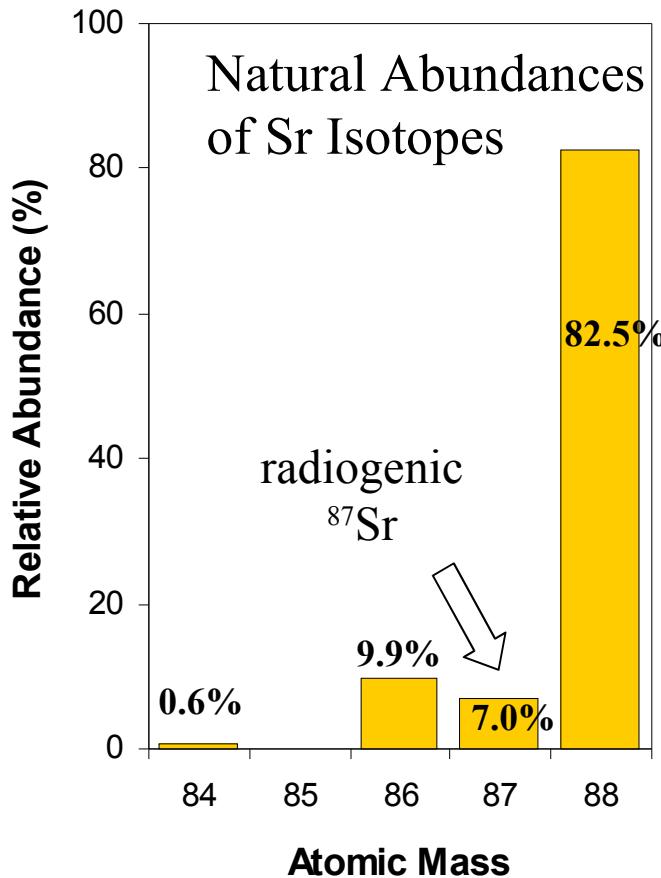


Objectives

- Quantify the proportion of weathering and atmospheric sources of Ca in tree nutrition by the Sr isotope method
- Study the evolution of stand nutrition by Sr isotope tree-ring analysis of Sr isotopes

Method

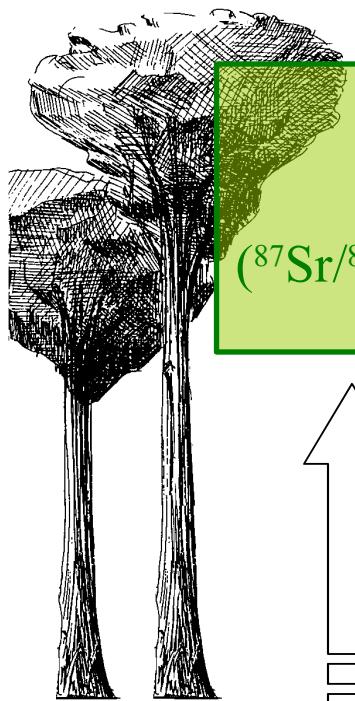
- Ca and Sr have a similar geochemical behaviour
- Similar ionic radius : Ca = 1.00 Å ; Sr = 1.13 Å
- Identical charge (+2)



$^{87}\text{Sr}/^{86}\text{Sr}$ ratio

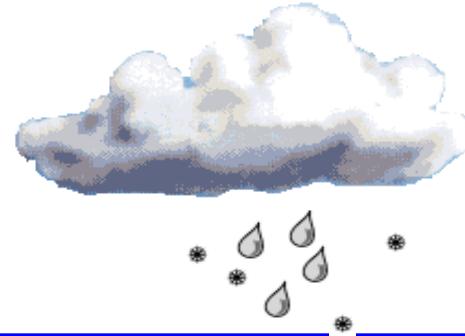
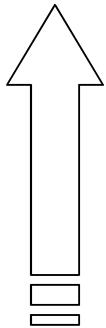
Method

The use of strontium isotope ratio



Weathering
(source 1)
 $(^{87}\text{Sr}/^{86}\text{Sr})_{Wea} = \text{high}$
 $\text{Sr/Ca} = \sim 0.0060$

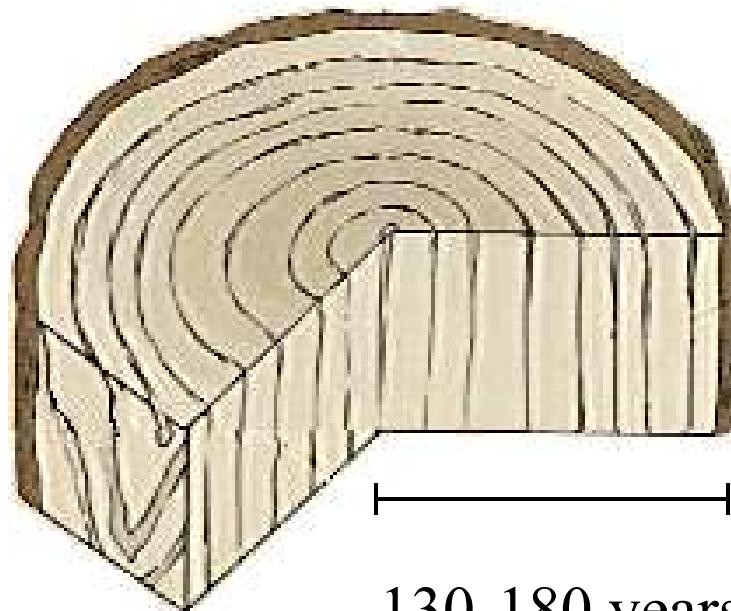
Vegetation
(mixing)
 $(^{87}\text{Sr}/^{86}\text{Sr})_{Mix} = \text{intermediate}$



Atmospheric precipitation
(source 2)
 $(^{87}\text{Sr}/^{86}\text{Sr})_{Prec} = 0.709 \text{ (low)}$
 $\text{Sr/Ca} = \sim 0.0030$

$$\%(\text{Ca})_{Prec} = \frac{[(^{87}\text{Sr}/^{86}\text{Sr})_{Mix} - (^{87}\text{Sr}/^{86}\text{Sr})_{Wea}] (\text{Sr/Ca})_{Wea}}{[(^{87}\text{Sr}/^{86}\text{Sr})_{Mix} - (^{87}\text{Sr}/^{86}\text{Sr})_{Wea}] (\text{Sr/Ca})_{Wea} - [(^{87}\text{Sr}/^{86}\text{Sr})_{Prec} - (^{87}\text{Sr}/^{86}\text{Sr})_{Mix}] (\text{Sr/Ca})_{Prec}} \times 100$$

Sr isotope dendrochemistry

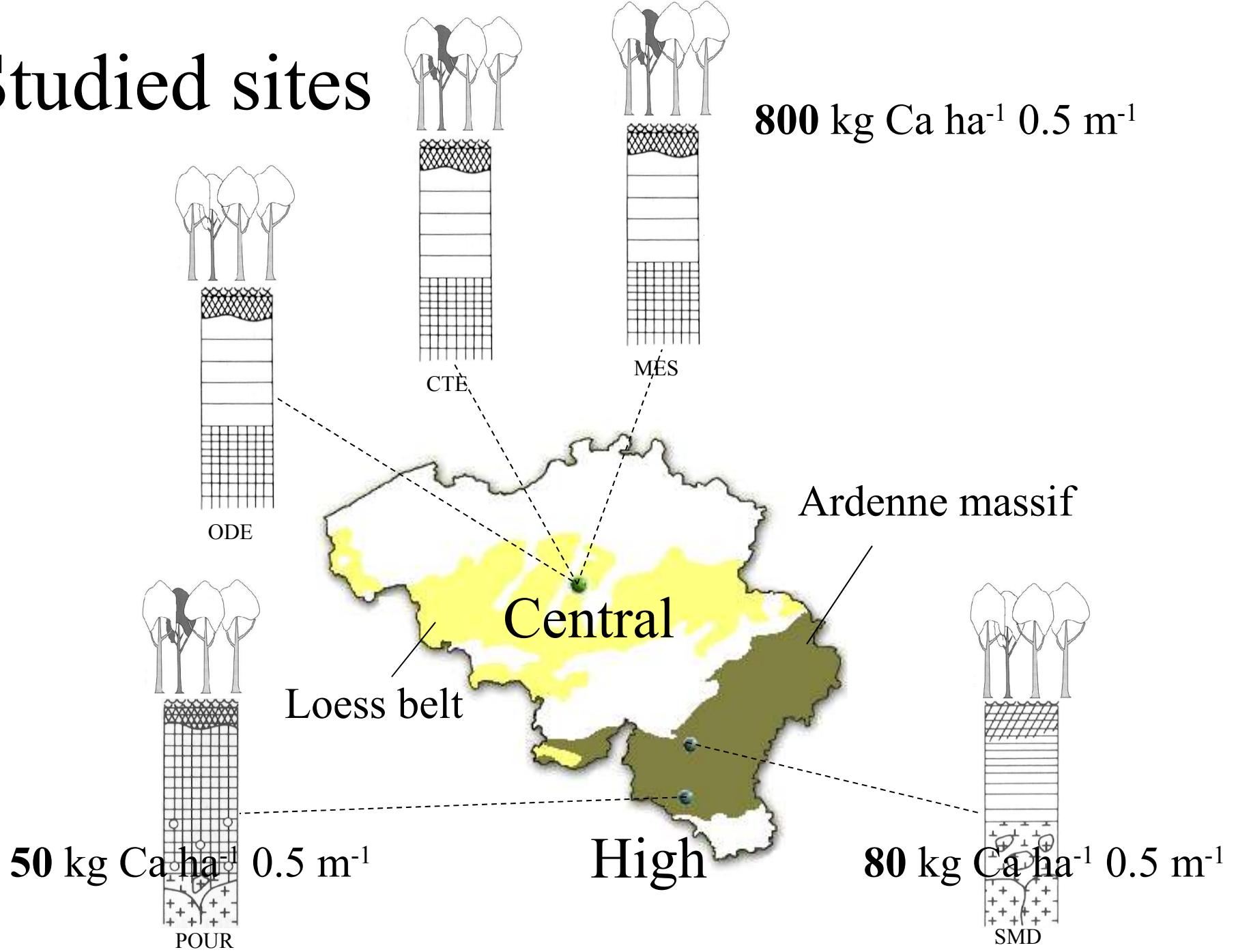


130-180 years

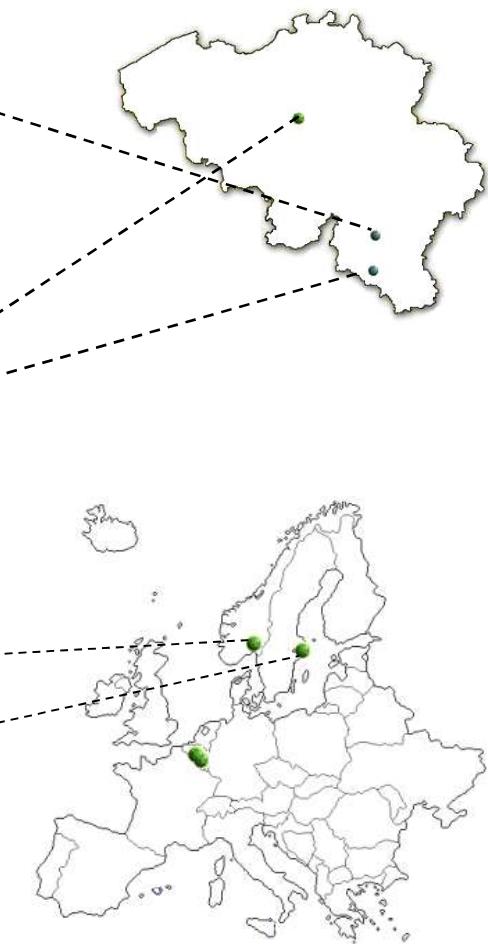
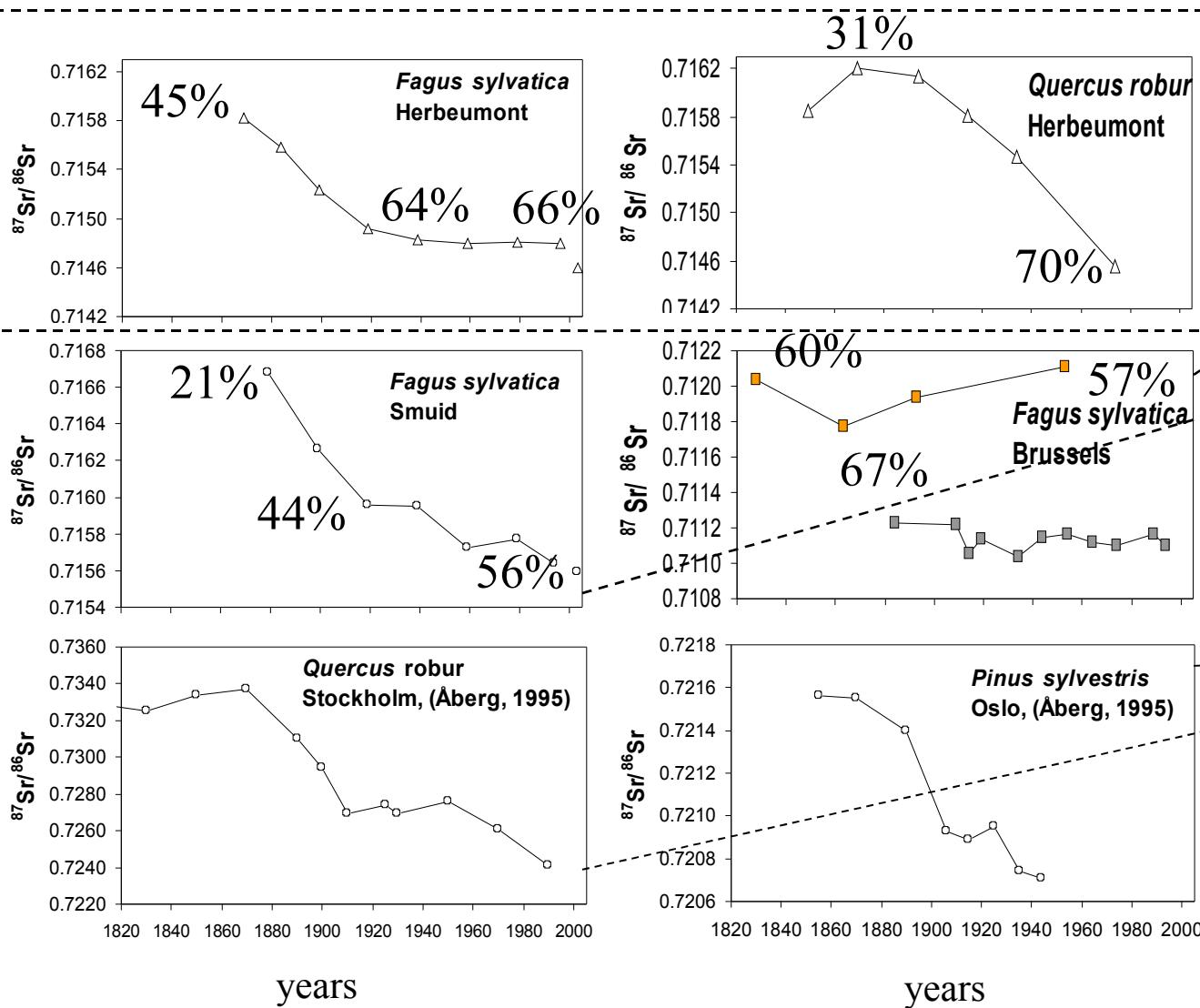
Fagus sylvatica L.

Quercus robur L.

Studied sites

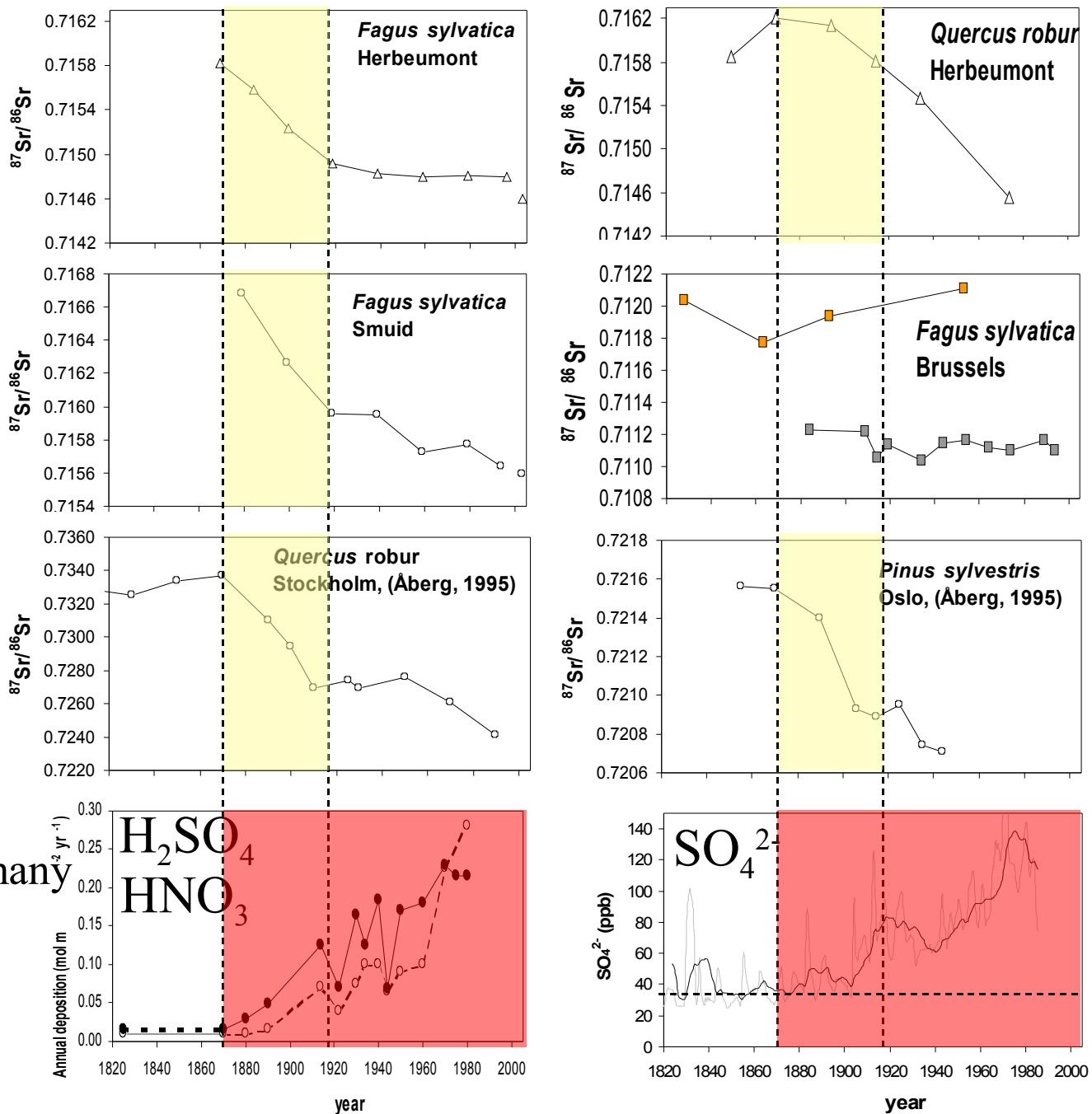


Evolution of the atmospheric source contribution to tree Ca nutrition



Mean increase of the atmospheric Ca of 20-30%

Evolution of the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio



Acid deposition, Germany
Ulrich, 1987

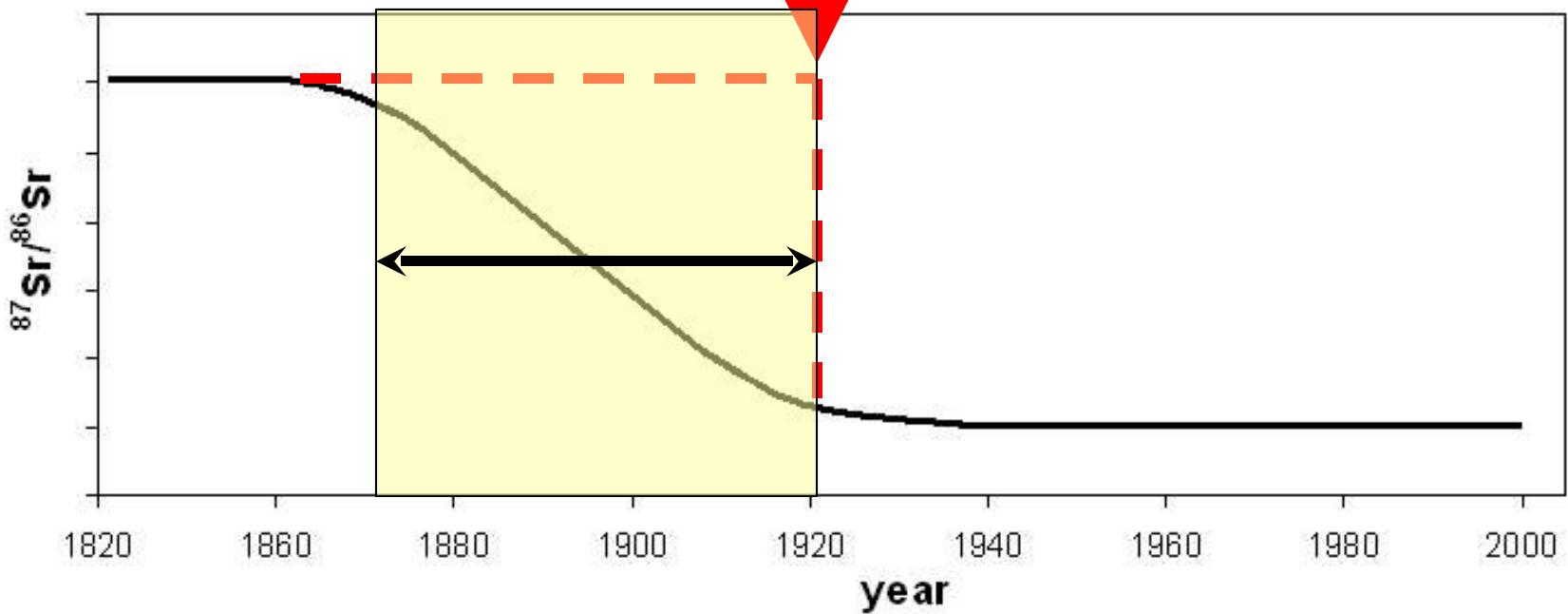
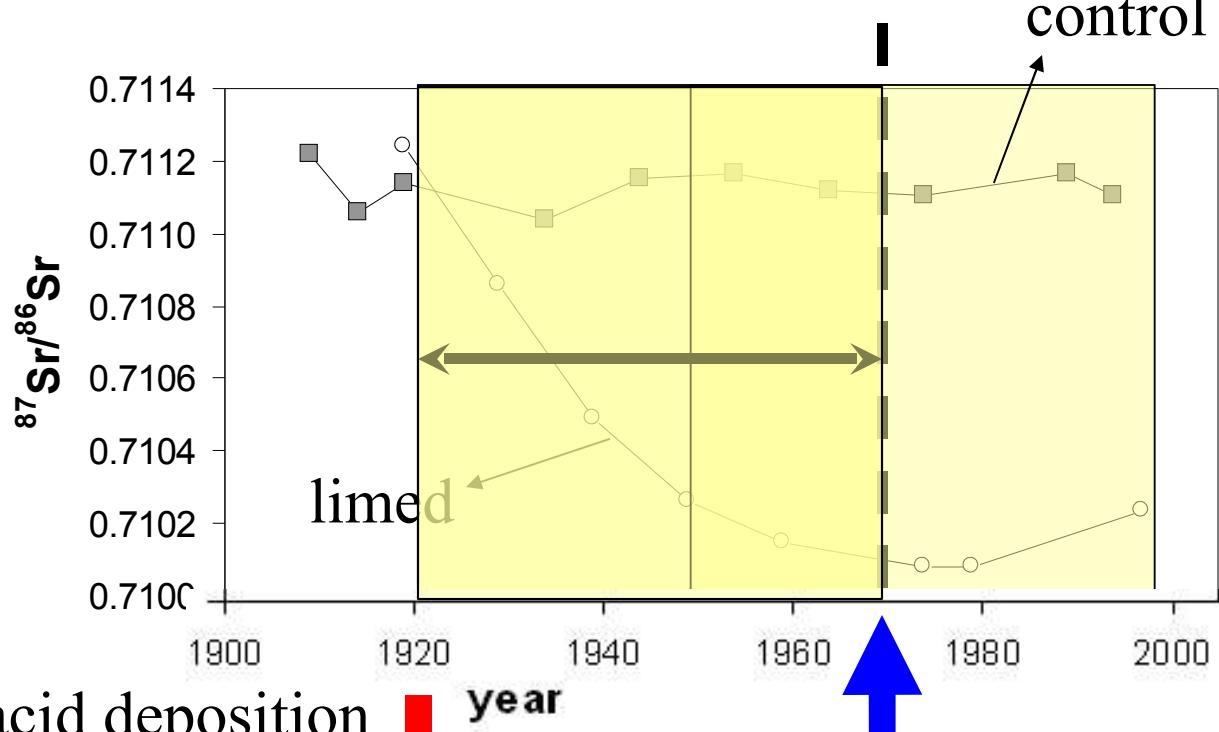
GISP2, Greenland
Mayewski *et al.*, 1990

Limed stand

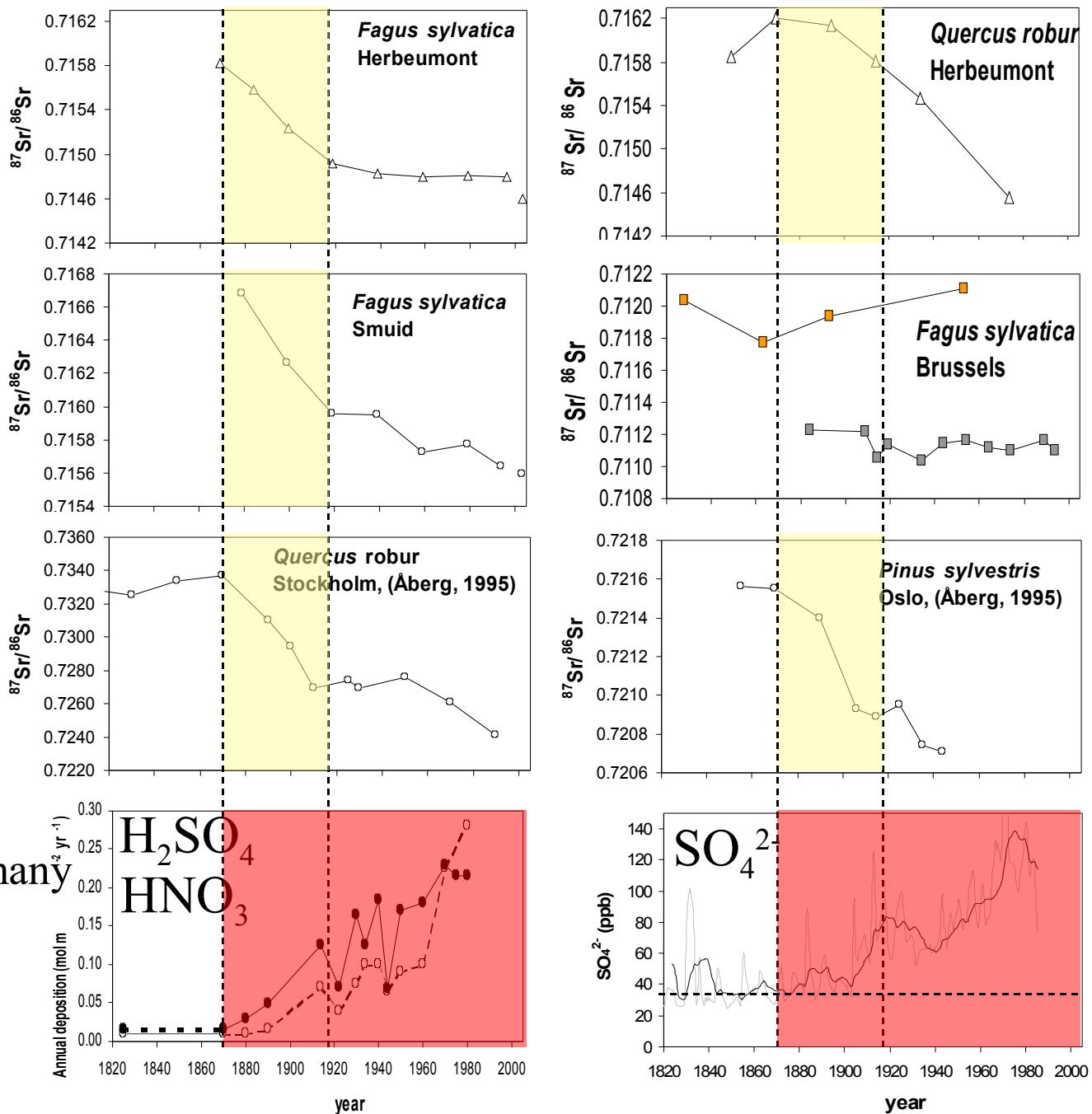
4 t of CaCO₃

$$^{87}\text{Sr}/^{86}\text{Sr} = 0.707857$$

schematic curve
of stands depleted by acid deposition



Evolution of the $^{87}\text{Sr}/^{86}\text{Sr}$ ratio

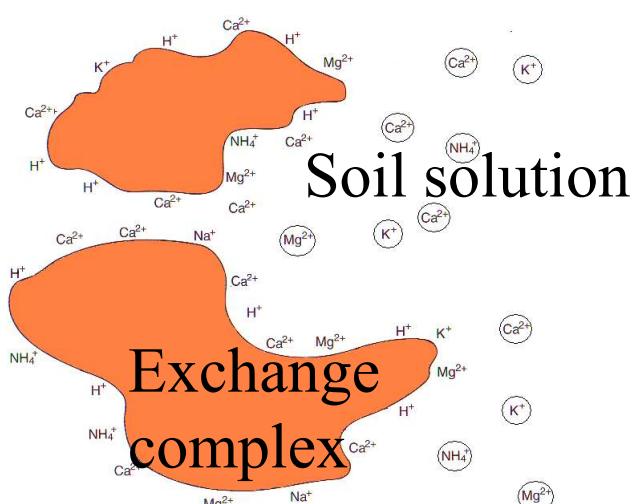


Mechanism

Pristine forest soil



Weak acid



Tree-rings

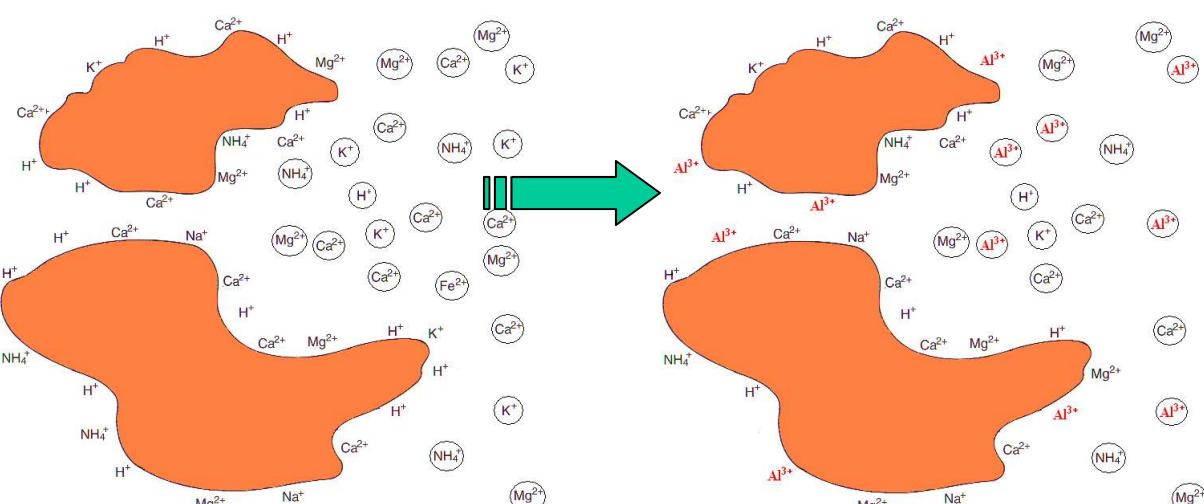


Post-industrial forest soil



Strong acid

Tomlinson, 2003



New equilibrium

Conclusion

- Continuing atmospheric inputs of strong acids can change the source of Ca for tree nutrition
- Probably linked to a change in soil equilibrium reactions
- The sulphur pollution abatement measures are not sufficient to return to pristine soil functioning and tree nutrition.